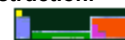


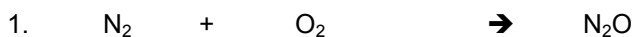
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HONORS WORKSHEET 5d: Reaction Types

Consider each of the following UNBALANCED equations. Identify the TYPE of reaction that is taking place by choosing from the following list. In each case choose the number of types shown in parentheses to classify the reaction.

- REDOX
- single displacement
- double displacement
- acid-base
- precipitation
- metal displacement
- gas producing
- disproportionation
- halogen displacement
- synthesis (combination)
- decomposition
- combustion
- hydrogen displacement (from acid or water)



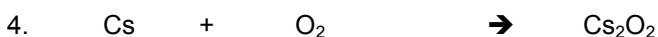
(2 types):



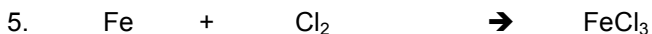
(2 types):



(2 types):



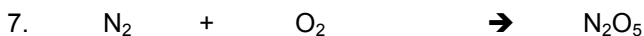
(2 types):



(2 types):

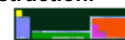


(2 types):

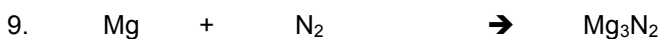


(2 types):

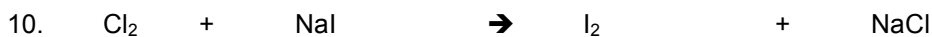
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(2 types):



(2 types):



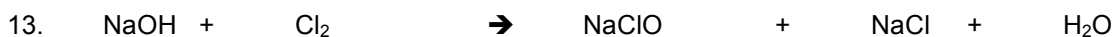
(2 types):



(1 type):



(3 types):



(2 types):



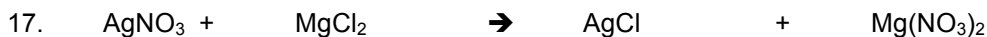
(2 types):



(2 types):

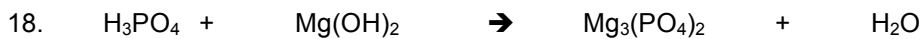
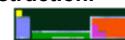


(2 types):



(2 types):

Revised August 2012



(2 types):



(2 types):



(2 types):



(2 types):



(2 types):



(2 types):



(2 types):



(3 types):